

# UNITED STATES PATENT OFFICE.

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## MANUFACTURE OF OXIDS OF NITROGEN.

1,207,706.

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No Drawing.

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*To all whom it may concern:*

Be it known that we, CARL BOSCH, ALWIN MITTASCH, and CHRISTOPH BECK, citizens of the German Empire, residing at Ludwigs-hafen-on-the-Rhine, Germany, have in-vented new and useful Improvements in the Manufacture of Oxids of Nitrogen, of which the following is a specification.

We have found that the catalytic oxida-tion of ammonia with the formation of oxids of nitrogen is effected in a very advantageous manner by passing a mixture of ammonia and oxygen-containing gas, such as air or oxygen, over a hot catalytic agent contain-  
ing at least one oxid of a metal of the iron group (in particular an oxid of iron, manga-nese, chromium or uranium) and at least one oxid of a rare earth metal, and that very favorable results are obtained, if the catalytic agents contain, besides said con-stituents a compound of bismuth, for exam-ple an oxid of bismuth.

In addition to the above compounds, the catalytic agent may contain other substances, for instance, binding agents, such as cal-cium oxid, magnesia or alumina, but it is preferred to avoid the presence of certain non-metallic elements and metalloids and compounds thereof, such as sulfuric acid, phosphoric acid, boric acid and silica, or at any rate larger quantities thereof.

The production of the catalytic mixtures can advantageously be carried out by pre-cipitating or calcining mixtures of salts of the components and then forming into suit-ably shaped pieces or lumps; and we prefer to employ the said catalytic mixtures in a layer, or layers, of small lumps. The pro-  
portions of the ingredients can be varied considerably.

The following example will serve to fur-ther illustrate the nature of our invention which, however, is not confined to the exam-ple. The parts are by weight.

Example: Dissolve 15 parts of ferric ni-trate and 1 part of lanthanum nitrate in water and precipitate the hot solution with ammonia. Filter, wash well, form into suitably shaped lumps, dry at 250° C., place the mass in a contact tube and then heat at

about 600° C., then pass a mixture of am-monia and air through the mass at about 750° C. In this example instead of lantha-num nitrate, other rare earth metals, or mix-tures thereof, in particular the cerite and the yttria earths, can be added and further salts or oxids of another metal of the iron group or of two, or more, of those metals can be employed. Thus, for instance, catalytic mixtures containing ferric oxid and yttrium oxid, or manganese oxid and cerium oxid, or manganese oxid, ferric oxid and didymium oxid may be used. In all cases a compound of bismuth may also be present.

Now what we claim is:—

1. The manufacture of oxids of nitrogen by passing a mixture of ammonia and an oxygen-containing gas over a heated cata-lytic agent containing at least one oxid of a metal of the iron group and at least one oxid of a rare earth metal.

2. The manufacture of oxids of nitrogen by passing a mixture of ammonia and an oxygen-containing gas over a heated cata-lytic agent containing at least one oxid of a metal of the iron group and at least one oxid of a rare earth metal and a compound of bismuth.

3. The manufacture of oxids of nitrogen by passing a mixture of ammonia and an oxygen-containing gas over a heated cata-lytic agent containing at least one oxid of a metal of the iron group and at least one oxid of a rare earth metal and an oxid of bismuth.

4. The manufacture of oxids of nitrogen by passing a mixture of ammonia and an oxygen-containing gas through a layer of separate lumps of a heated catalytic agent containing at least one oxid of a metal of the iron group and at least one oxid of a rare earth metal.

5. The manufacture of oxids of nitrogen by passing a mixture of ammonia and an oxygen-containing gas through a layer of separate lumps of a heated catalytic agent containing at least one oxid of a metal of the iron group and at least one oxid of a rare earth metal and a compound of bis-muth.

6. The manufacture of oxids of nitrogen  
by passing a mixture of ammonia and an  
oxygen-containing gas through a layer of  
separate lumps of a heated catalytic agent  
5 containing at least one oxid of a metal of  
the iron group and at least one oxid of a  
rare earth metal and an oxid of bismuth.  
In testimony whereof we have hereunto

set our hands in the presence of two sub-  
scribing witnesses.

CARL BOSCH.  
ALWIN MITTASCH.  
CHRISTOPH BECK.

Witnesses:

ARTHUR DENONVILLE,  
JOHANNES ACHMEL.